Buffer

The use of cacodylate buffer $(Na(CH_3)2AsO_2)$ avoids the presence of extraneous phosphates which may precipitate out in the later staining procedure with uranyl acetate. The buffer also has a good range within which the pH is stable. Be careful: The buffer is not volatile and can be opened out of the fume hood but it contains arsenic so should not be spilled or imbibed.

Amino acid-containing buffers are not used with glutaraldehyde since they can be expected to react with aldehyde. Phosphate buffers are used for immuno labelling as cacodylate buffer may affect the antibody.

- 1. Stock solution is 0.2M sodium cacodylate (MW 214.02)
- 2. Dissolve 3.29 g in 100 mls ddw. Adjust pH with HCl